

Nicholas Reactions in the Synthesis of Dicobalt Dibenzocyclooctyne Complexes

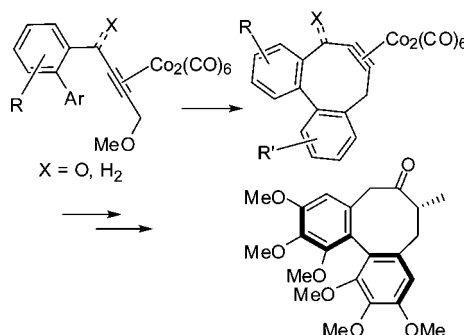
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Received September 10, 2013

ABSTRACT



Hexacarbonyldicobalt complexes of biaryl-substituted 4-methoxybutynones and 4-methoxy-2-butyne undergo intramolecular Nicholas reactions to form dibenzocyclooctyne–Co₂(CO)₆ complexes in good yields. Reductive decomplexation of the cyclization products is possible, and the method has been applied to a formal synthesis of isoschizandrin.

Cyclooctyne is well-known as the smallest of the simple cycloalkynes with sufficient stability to be capable of isolation in the conventional sense.¹ This does not apply to all cyclooctyne derivatives, as increasing unsaturation in the eight-membered ring renders the compounds more marginally stable² or incapable of isolation.³ In contrast, the hexacarbonyldicobalt complexes of cyclooctynes appear to have excellent stability. While direct preparation from cyclooctyne itself is known,⁴ this is synthetically limited. Several scattered reports of de novo construction of cyclooctyne–Co₂(CO)₆ complexes have been published,

including those resulting from Nicholas reaction chemistry,⁵ ring-closing metathesis,⁶ aldol and Michael reaction chemistry,⁷ Diels–Alder reactions,⁸ and epoxide ring-openings.⁹ In addition, cyclic ether and amine complexes have been prepared.¹⁰ Despite the viability of systems of this class, there has been no attempt to prepare dibenzocyclooctynedicobalt complexes (**1**, Figure 1) or to explore their applicability toward dibenzocyclooctane-containing compounds.

The dibenzocyclooctane lignans are a large group of natural products occurring widely, particularly in the Schizandraceae family.¹¹ Their structural features and

(1) For reviews, see: (a) Krebs, A.; Wilke, J. *Top. Curr. Chem.* **1983**, 109, 189. (b) Hopf, H.; Grunenberg, J. In *Strained Hydrocarbons. Beyond the van't Hoff and Le Bel Hypothesis*; Dodziuk, H., Ed.; Wiley-VCH: Weinheim, 2009; pp 375–397.

(2) (a) Varga, B. R.; Kallay, M.; Hegyi, K.; Beni, S.; Kele, P. *Chem.—Eur. J.* **2012**, 18, 822. (b) Krebs, A.; Oldenthal, J.; Kimling, H. *Tetrahedron Lett.* **1976**, 4663. (c) Werner, C.; Hopf, H.; Grunenberg, J.; Jones, P. G. *Eur. J. Org. Chem.* **2010**, 4027.

(3) (a) Meier, H.; Layer, A.; Zetzsche, A. *Chem. Ztg.* **1974**, 98, 460. (b) Sletten, E. M.; Nakamura, H.; Jewett, J. C.; Bertozzi, C. R. *J. Am. Chem. Soc.* **2010**, 132, 11799. (c) Gugel, H.; Meier, H. *Chem. Ber.* **1981**, 113, 1431.

(4) (a) Bennett, M. A.; Donaldson, P. B. *Inorg. Chem.* **1978**, 17, 1995. (b) Petersen, H.; Meier, H. *Nouv. J. Chim.* **1980**, 4, 687.

(5) (a) Jamison, T. F.; Shambayati, S.; Crowe, W. E.; Schreiber, S. L. *J. Am. Chem. Soc.* **1994**, 116, 5505. (b) Atkinson, R. C. J.; Hope-Weeks, L. J.; Mays, M. J.; Solan, G. A. *J. Organomet. Chem.* **2007**, 692, 2076.

(6) (a) Green, J. R. *Synlett* **2001**, 353. (b) Young, D. G. J.; Burlison, J. A.; Peters, U. *J. Org. Chem.* **2003**, 68, 3494.

(7) (a) Mitachi, K.; Shimizu, T.; Miyashita, M.; Tanino, K. *Tetrahedron Lett.* **2010**, 51, 3983. (b) Inaba, K.; Takaya, J.; Iwasawa, N. *Chem. Lett.* **2007**, 474.

(8) Iwasawa, N.; Inaba, K.; Nakayama, S.; Aoki, M. *Angew. Chem., Int. Ed.* **2005**, 44, 7447.

(9) Nagumo, S.; Ishii, Y.; Sato, G.; Mizukami, M.; Imai, M.; Kawahara, N.; Akita, H. *Tetrahedron Lett.* **2009**, 50, 26.

(10) (a) Isobe, M.; Hamajima, A. *Nat. Prod. Rep.* **2010**, 27, 1204 and references therein. (b) Mizukami, M.; Saito, H.; Higuchi, T.; Imai, M.; Nade, H.; Kawahara, N.; Nagumo, S. *Tetrahedron Lett.* **2007**, 48, 7228. (c) Closser, K. D.; Quintal, M. M.; Shea, K. M. *Org. Lett.* **2009**, 11, 3680. (d) Mukai, C.; Kojima, T.; Kawamura, T.; Inagaki, F. *Tetrahedron* **2013**, 69, 7659.

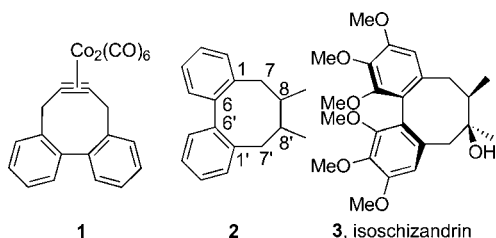


Figure 1. Dibenzocyclooctane lignan framework and **1**.

wide-ranging biological activities have made them recent attractive synthetic targets. Synthesis of the eight-membered rings of these systems is overwhelmingly accomplished either by aryl–aryl (C6–C6') (see **2**) coupling protocols^{11a,12} or by assorted condensation or coupling reactions at the homobenzylic sites (C8–C8') of functionalized 2,2'-diethylbiphenyls. Construction by way of benzylic–bishomobenzylic bond formation (C8'–C7') of biaryls is less common but known.¹³ Their preparation by way of C1'–C7' coupling reactions, such as by electrophilic substitution protocols, is rare and on those occasions tend to be by dienone–phenol rearrangements.^{14,15}

Our group has had recent success with the use of intramolecular Nicholas reaction chemistry in the preparation of dibenzocycloheptyne–Co₂(CO)₆ complexes¹⁶ and have found the method useful in allocolchicine synthesis in conjunction with reductive decomplexation reactions. As a result of these developments, we have chosen to explore a Nicholas reaction approach to such dibenzocyclooctyne complexes, with a view toward their use in dibenzocyclooctane synthesis. Isoschizandrin (**3**), an antiulcer C-8 oxygenated dibenzocyclooctane lignan, was identified as a target compound relevant to this chemistry.¹⁷ Given the common occurrence of C7-oxygen substituted dibenzocyclooctane lignans in addition to their C8-hydroxy-substituted and nonoxygen-substituted counterparts,^{11a} we considered it of importance to include both γ -carbonyl cation (**4**→**5**) and normal (**6**→**7**) versions of these Nicholas reactions.

(11) (a) Jiang, C.; Reiner, J.; Xie, J. *Chem. Rev.* **2005**, *105*, 4581. (b) See also: Bringmann, G.; Gulder, T.; Gulder, T. A. M.; Breuning, M. *Chem. Rev.* **2011**, *111*, 563. (c) Sefkow, M. *Top. Curr. Chem.* **2005**, *243*, 185.

(12) For a recent example, see: Zheng, S.; Aves, S. J.; Laraia, L.; Galloway, W. R. J. D.; Pike, K. G.; Wu, W.; Spring, D. R. *Chem.—Eur. J.* **2012**, *18*, 3193.

(13) (a) Li, Y.; Wang, Q.; Dong, L.; Guo, X.; Wang, W.; Xie, J.; Chang, J. *Synthesis* **2009**, 3383. (b) Monovich, L. G.; Huerou, Y. L.; Rönn, M.; Molander, G. A. *J. Am. Chem. Soc.* **2000**, *122*, 52. (c) Robin, J.-P.; Dhal, R.; Brown, E. *Tetrahedron* **1984**, *40*, 3509.

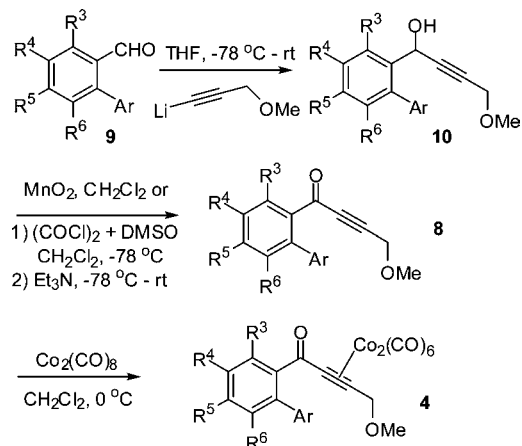
(14) (a) Bowden, B. F.; Read, R. W.; Taylor, W. C. *Aust. J. Chem.* **1981**, *34*, 799. (b) Pelter, A.; Ward, R. S.; Abd-El-Ghani, A. *J. Chem. Soc., Perkin Trans. I* **1992**, 2249.

(15) For a rare, low-yielding, exception, see: Plummer, E. L.; Seiders, R. A. H.; Seelye, D. E.; Stewart, R. R. *Pestic. Sci.* **1984**, *15*, 509.

(16) (a) Djurdjevic, S.; Yang, F.; Green, J. R. *J. Org. Chem.* **2010**, *75*, 8241. (b) Djurdjevic, S.; Green, J. R. *Org. Lett.* **2007**, *9*, 5505.

(17) (a) Warshawsky, A. M.; Meyers, A. I. *J. Am. Chem. Soc.* **1990**, *112*, 8090. (b) Tanaka, M.; Mukaiyama, C.; Mitsuhashi, H.; Maruno, M.; Wakamatsu, T. *J. Org. Chem.* **1995**, *60*, 4339. (c) Molander, G. A.; George, K. M.; Monovich, L. G. *J. Org. Chem.* **2003**, *68*, 9533. (d) Isolation: Ikeya, Y.; Taguchi, H.; Mitsuhashi, H.; Takeda, S.; Kase, Y.; Aburada, M. *Phytochemistry* **1988**, *27*, 569.

Table 1. Preparation of **4**



9	10	8	4
9a , R ⁴ = OMe, Ar = 2,3,4-(MeO) ₃ C ₆ H ₂	10a (86%)	8a (96%) ^a	4a (93%)
9b , R ³ = R ⁴ = OMe, Ar = 2,3,4-(MeO) ₃ C ₆ H ₂	10b (97%)	8b (78%) ^b	4b (90%)
9c , R ⁴ = OMe, Ar = 3-thienyl	10c (94%)	8c (79%) ^b	4c (87%)
9d , R ⁴ = R ⁵ = R ⁶ = OMe, Ar = 2,3,4-(MeO) ₃ C ₆ H ₂	10d (98%)	8d (92%) ^a	4d (93%)

^a MnO₂, CH₂Cl₂, rt; ^b Swern conditions.

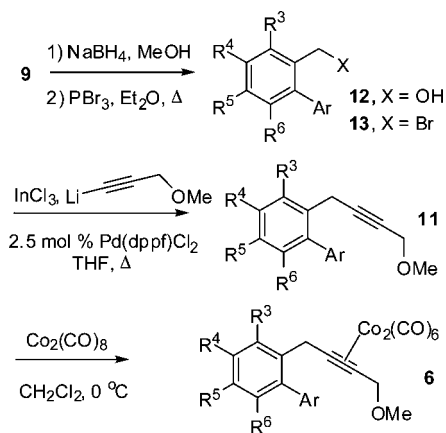
The precursors for γ -carbonyl cation complexes were selected to be 4-alkoxy-2-butynyl-substituted biaryls (**8**) (Table 1), which were prepared from the biarylcarboxaldehydes (**9**)¹⁶ in straightforward fashion.

Reaction of the aldehydes with the lithium acetylide derived from 3-methoxy-1-propyne (propargyl methyl ether) gave the benzylic/propargylic alcohols (**10**) in good to excellent yield (Table 1); subsequent oxidation with MnO₂, or using Swern conditions when MnO₂ performed sluggishly, gave the corresponding ketones (**8**). Complexation of the alkyne functions of these alkynones with Co₂(CO)₈ then afforded **4** readily.

The biaryls bearing 4-methoxy-2-butynyl functions (**11**) were also prepared from the biarylcarboxaldehydes (**9**) in three steps (Table 2). Reduction of the aldehyde function to the benzylic alcohols (**12**) occurred cleanly and in excellent yields. Substitution of bromide for the alcohol function (**13**) was accomplished with PBr₃. For tetramethoxy-substituted **13f**, reaction of the benzyl bromide with the lithium acetylide derived from propargyl methyl ether afforded **11f** in acceptable yield. In other cases, this protocol gave poor yields; conversely, use of this lithium acetylide in the presence of InCl₃ and catalytic amounts of Pd(dppf)Cl₂ gave **11c–e** successfully.¹⁸ Once again, the alkyne functions underwent complexation by Co₂(CO)₈ to afford **6** readily.

Cyclization reactions of the aryl alkynone complexes were investigated first. While previous experience has

(18) Pérez, I.; Sestelo, J. P.; Sarandeses, L. A. *J. Am. Chem. Soc.* **2001**, *123*, 4155.

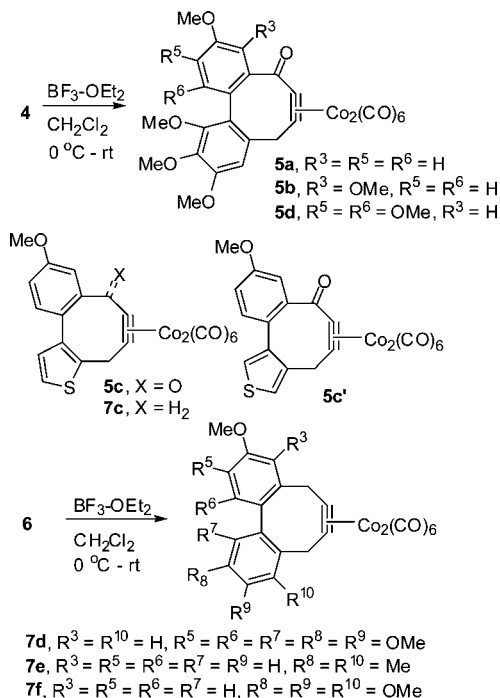
Table 2. Preparation of **6**


9	12	13	11	6
9c , R ⁴ = OMe, Ar = 3-thienyl	12c (89%)	13c (73%)	11c (61%)	6c (86%)
9d , R ⁴ = R ⁵ = R ⁶ = OMe, Ar = 2,3,4-(MeO) ₃ C ₆ H ₂	12d (93%)	13d (99%)	11d (69%)	6d (99%)
9e , R ⁴ = OMe, Ar = 3,5-Me ₂ C ₆ H ₃	12e (92%)	13e (63%)	11e (66%)	6e (86%)
9f , R ⁴ = OMe, Ar = 3,4,5-(MeO) ₃ C ₆ H ₂	12f (99%)	13f (80%)	11f (61%) ^a	6f (78%)

^a InCl₃ and Pd(dppf)Cl₂ omitted.

shown that Nicholas reaction based γ -carbonyl cations are more reliably generated using Bu₂BOTf as Lewis acid,¹⁹ BF₃·OEt₂ (3 equiv, 0 °C) gave good rates of reaction in the case of **4** (Table 3). Reactions were conducted at 4 × 10⁻³ M; doubling the concentration reduced yield modestly (entry 4 versus entry 5). The addition of *i*-Pr₂NEt (1.5 equiv, with 4 equiv BF₃·OEt₂) occasionally resulted in lesser amounts of decomposition due to presumed scavenging of liberated acid and consequently gave greater yields. Ultimately, tetramethoxy **4a** afforded **5a** in 85% yield, whereas hexamethoxy substrate **4d** gave **5d** in 81% yield. In the case of pentamethoxy substrate **4b** and thiophene-containing **4c**, the reactions were conducted in the presence of *i*-Pr₂NEt; the former afforded **5b** in 71% yield, whereas the latter gave **5c** in 68% yield, as a 14:1 mixture of products reacting at C-2 and C-4 (**5c'**) of the thiophene ring.

Cyclization reactions involving the benzyl alkyne complexes **6** succeeded under similar conditions. In none of the cases was the presence of additional *i*-Pr₂NEt necessary, and the reactions were somewhat more rapid than for **4**. We attributed this to the lack of a competitively Lewis basic and electron-withdrawing carbonyl in **6**. In the event, hexamethoxy-substituted **6d** gave dibenzocyclooctyne **7d** in 93% yield in 1 h, whereas tetramethoxy-substituted **6f** afforded **7f** in 91% yield over the same period. The substrates with less electron rich arene nucleophiles also underwent cyclization rapidly, as thiophene-substituted **6c** afforded **7c** in 77% yield over 2 h, while dimethyl-substituted **6e** gave **7e** in 88% over 2 h. In the **6c**→**7c** case,

Table 3. Intramolecular Nicholas Reactions


entry	starting material	conditions ^a	time (h)	product	yield (%)
1	4a	A	5	5a	85
2	4b	B	6	5b	71
3	4c	B	8	5c	68 ^b
4	4d	A	8	5d	81
5	4d	A ^c	8	5d	71
6	6c	A	1	7c	77
7	6d	A	2	7d	93
8	6e	A	2	7e	88
9	6f	A	1	7f	91

^a A: BF₃·OEt₂ (3 equiv), 0 °C to rt, CH₂Cl₂ (4 × 10⁻³ M); B: BF₃·OEt₂ (4 equiv), *i*-Pr₂NEt (1.5 equiv) 0 °C to rt, CH₂Cl₂ (4 × 10⁻³ M). ^b **5c**:**5c'** = 14:1. ^c 8 × 10⁻³ M.

there was no evidence of C-4 reactivity on the thiophene ring competing with the C-2 substitution.

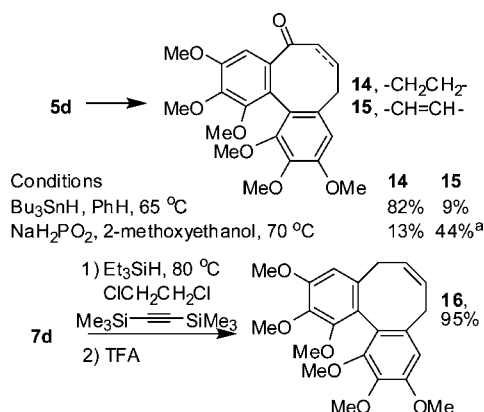
Decomplexation reactions of the cyclooctynones were studied using **5d** as a model compound (Scheme 1). Use of Bu₃SnH²⁰ resulted in the successful removal of the Co₂(CO)₆ unit with predominant overreduction of the alkyne function to give cyclooctanone **14** (82% yield), along with a small amount of cyclooctenone **15** (9% yield). The cyclooctenone **15** could be obtained as the predominant product (44% yield, 51% based on recovered starting material) by employing 2 equiv of NaH₂PO₂ in 2-methoxyethanol;²¹ this was accompanied by 13% of cyclooctanone **14** and 14% of unreacted **5d**. The use of the conventionally employed 5 equiv of hypophosphite gave greater amounts of cyclooctanone **14** (29%), at the expense of **15** (36%).²²

(20) Hosokawa, S.; Isobe, M. *Tetrahedron Lett.* **1998**, 39, 2609.

(21) Takai, S.; Ploypradith, P.; Hamajima, A.; Kira, K.; Isobe, M. *Synlett* **2002**, 588.

(19) Jacobi, P. A.; Buddhu, S. C.; Fry, D.; Rajeswari, S. *J. Org. Chem.* **1997**, 62, 2894.

Scheme 1. Reductive Decomplexations



^a 51% based on recovered starting material.

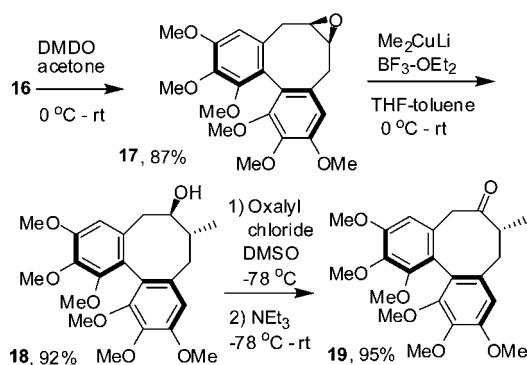
In the case of dibenzocyclooctyne complex **7d**, the reductive decomplexation was much more straightforward. Employing our hydrosilylation–protodesilylation modification of the Isobe protocol,^{16,23} **7d** afforded **16** cleanly (95% yield).

Alkene **16** is well suited for use in the synthesis of isoschizandrin. Epoxidation of the alkene function occurred readily with dimethyldioxirane (DMDO), giving **17** in 87% yield (Scheme 2). Lewis acid mediated cuprate attack of the epoxide gave alcohol **18** with complete diastereoselectivity (92% yield). Swern oxidation of the alcohol then afforded **19** in 95% yield. The Meyers group has previously converted enantioenriched **19** into (–)-isoschizandrin (79% yield, along with 9%

(22) Use of Et_3SiH resulted in a regioisomeric mixture of vinylsilane-bearing cyclooctenones which resisted protodesilylation.

(23) Kira, K.; Tanda, H.; Hamajima, A.; Baba, T.; Takai, S.; Isobe, M. *Tetrahedron* **2002**, 58, 6485.

Scheme 2. Completion of Isoschizandrin Formal Synthesis



(–)-schizandrin) by methyllithium addition;^{17a} consequently, this constitutes a formal synthesis of racemic isoschizandrin.

In summary, we have found that intramolecular Nicholas reactions of both biaryl-4-methoxybutynednicobalt complexes and biaryl-4-methoxy-2-butyndenicobalt complexes afford the corresponding dibenzocyclooctyne– $\text{Co}_2(\text{CO})_6$ complexes in good yields. Reductive decomplexation of these cyclization products is possible, and the process may be applied to the formal synthesis of isoschizandrin.

Acknowledgment. We are grateful to NSERC (Canada), the Canada Foundation for Innovation, and Ontario Innovation Trust for support of this research.

Supporting Information Available. Experimental procedures and spectroscopic data for all new compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.